

# Bands and electronic devices

Electrons and holes in bands

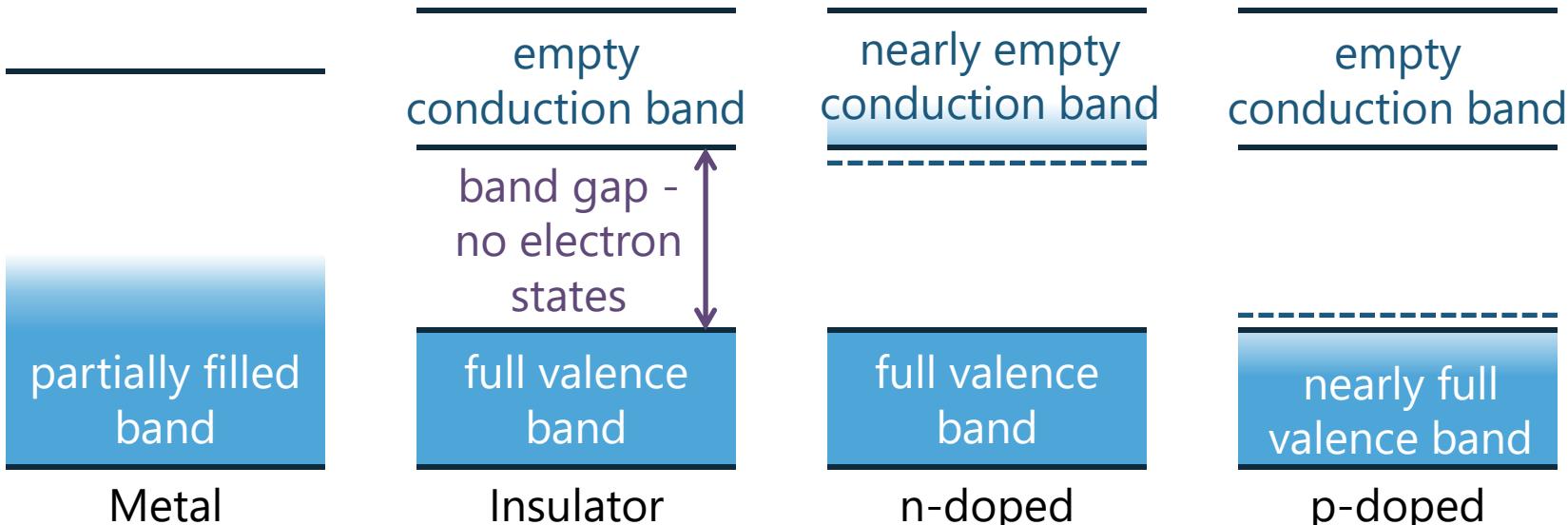
Modern physics for engineers

David Miller

# Metals, semiconductors and insulators - revision

# Metals, insulators and semiconductors - revision

Electron energy



electrons can move to new states  
**hence conducts electricity**

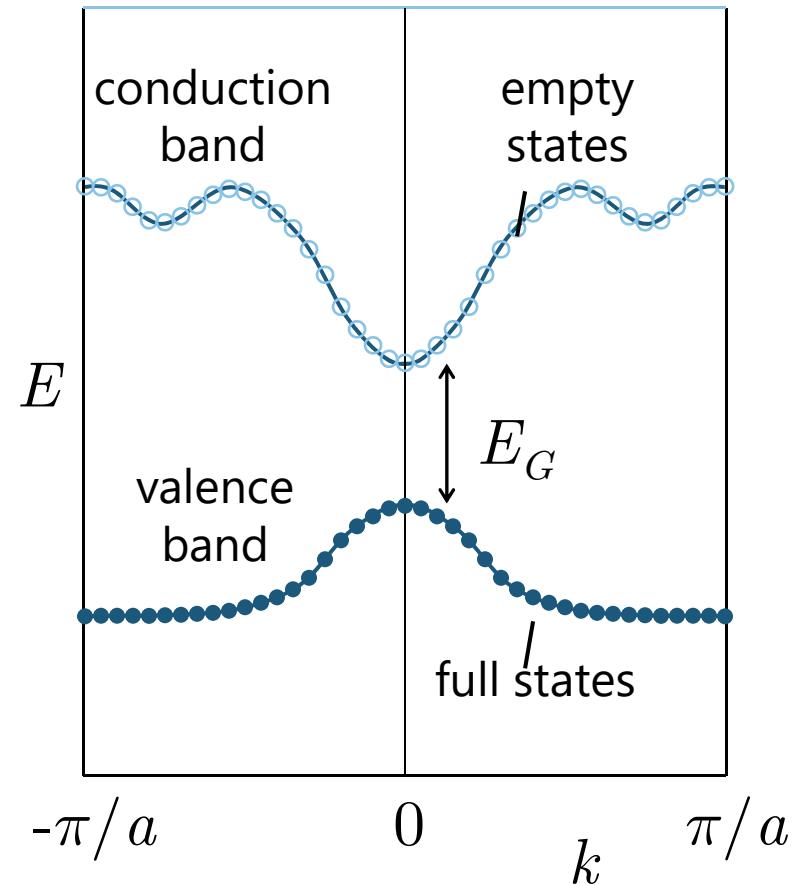
electrons in full bands cannot move to new states  
**does not conduct**

added free electrons in conduction band  
**conduct**

missing free electrons in valence band allow conduction

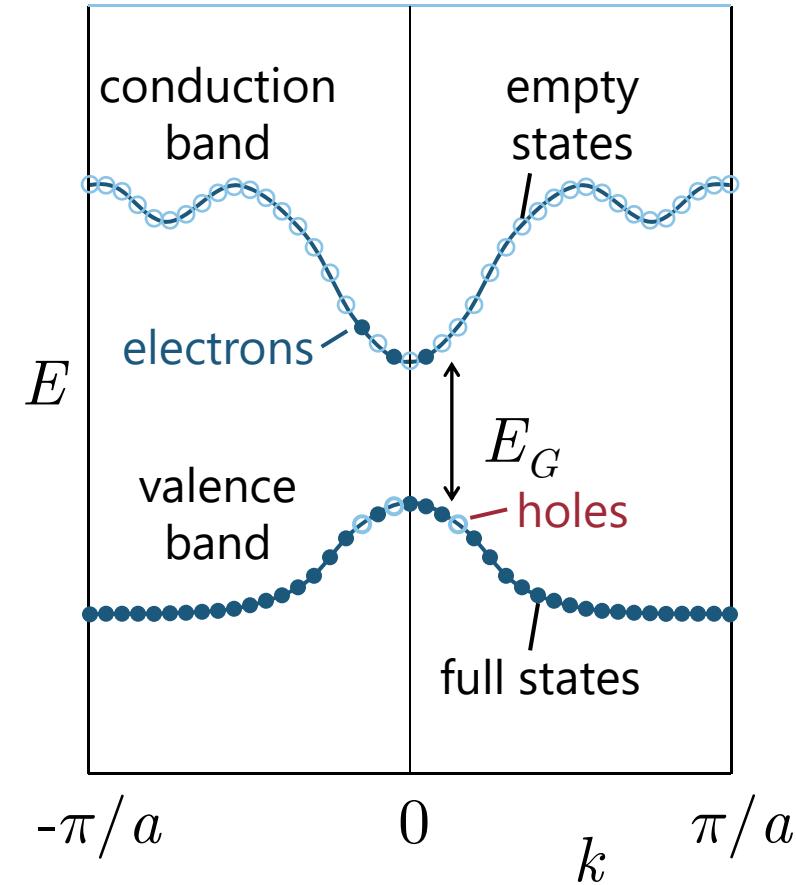
# Semiconductors and insulators - revision

Semiconductors and insulators have an (almost) completely full band the valence band separated by a “bandgap” energy  $E_G$  from an (almost) completely empty band the conduction band



# Semiconductors

Now we look at behaviors of electrons (and holes) in bands for example, for "transport" the movement of charge for conducting electrical current and some of the consequences of thermal distributions of them for devices



# Transport in semiconductors

# Transport of electrons

For an electron in some parabolic minimum (or maximum)

we can rewrite  $E = \frac{\hbar^2 k^2}{2m_{eff}}$

in terms of "crystal" momentum  $p_c = \hbar k$  as

$$E = \frac{p_c^2}{2m_{eff}}$$

Then  $\frac{dE}{dp_c} = \frac{p_c}{m_{eff}}$

# Transport of electrons

With 
$$\frac{dE}{dp_C} = \frac{p_C}{m_{eff}}$$

then thinking of a velocity  $v_g$  such that  $p_C = m_{eff}v_g$   
we have for this velocity at which we expect the  
particle is moving

$$v_g = \frac{dE}{dp_C}$$

This particular velocity  $v_g$   
is called the "group velocity"

# Transport of electrons

Suppose we apply an “external” force  $F$  to the particle  
such as from an electric field

Then the work done in applying the force through a  
distance  $dx$  is

$$dE = Fdx$$

The distance  $dx$  equals  
the group velocity  $v_g$  times  
the time,  $dt$ , for which the force is applied  
so we have

$$dE = Fdx = Fv_g dt$$

# Transport of electrons

Hence

$$F = \frac{1}{v_g} \frac{dE}{dt} = \frac{1}{v_g} \frac{dE}{dp_C} \frac{dp_C}{dt}$$

or, using  $v_g = dE / dp_C$

$$F = \frac{dp_C}{dt} \equiv \frac{d(\hbar k)}{dt}$$

So applying a force to an electron leads to  
“force is equal to rate of change of (crystal) momentum”

This crystal momentum behaves like the momentum of  
an effective particle of mass  $m_{eff}$

In this picture, applying a force  
moves the electron steadily through the Brillouin zone

# Ballistic transport of electrons



This kind of transport

where the electron is continuously  
accelerated by the applied field

is called “ballistic transport”

It does exist in materials like  
semiconductors

and is part of device analysis  
but it only applies for  
very short distances  
or very short time-scales

# Drift transport of electrons



More typically, the electron is accelerated ("ballistically") for some average or effective "scattering" time  $t_s$  then the electron is scattered by some collision with other electrons or crystal vibrations ("phonons") or crystal impurities or defects

# Drift transport of electrons



In the “drift” model

the scattering events are random  
but so strong that, on the average  
the electron velocity is  
randomized by them

So the average electron velocity after  
the collision is zero

because it could just as well be  
going in any direction

# Drift transport of electrons

Specifically, for an electric field of magnitude  $\mathcal{E}$

the magnitude of the force on an electron is  $F = e\mathcal{E}$

If we accelerate (from zero velocity)

using such a force for a time  $t_s$

the peak momentum just before scattering will be

$$p_C = \frac{dp_C}{dt} t_s = F t_s = e\mathcal{E} t_s = m_{eff} v_{peak}$$

The average velocity is half this peak value

$$v_{av} = \frac{et_s}{2m_{eff}} \mathcal{E}$$

# Drift transport of electrons

So, in drift transport, the average velocity of the electron  
is proportional to the applied electric field  $\mathcal{E}$   
and the average velocity is called the "drift velocity"

Such transport is often written as

$$v_{av} = \mu_e \mathcal{E}$$

where

$$\mu_e = \frac{e t_s}{2m_{eff}}$$

is called the "mobility"

This gives rise to behavior like Ohm's law  
with current proportional to the voltage

# Hole transport in semiconductors

# Holes



At a maximum at the top of the valence band

the electron effective mass is negative

Electrons would go backwards if pushed

Though counter-intuitive, this is correct

The group velocity can be “backwards”

e.g., in the “wrong” direction when pushed by an electric field

# Holes



For a set of electrons in states near such a negative effective mass maximum

but with a state (or wave packet of states) not occupied

that “empty” state or wave packet will move along

in the same “backwards” direction as the electrons do

# Holes



This “absence of an electron”  
wavepacket  
is moving in the wrong direction for  
electron electrical current  
but in the correct direction for the  
electrical current of a positively  
charged particle  
in this electric field

# Holes



So we can pretend this “absence-of-an-electron”

in a “negative electron effective mass” maximum

behaves like a positive charge with a positive mass

going in the “correct” direction

# Holes

This “cancellation of two minus signs”  
exchanging an absence of negative  
charge with a negative mass  
for an effective positive charge with  
a positive mass (of the same  
magnitude)  
lets us define the idea of a “hole”  
with positive charge and mass  
Hole transport otherwise obeys the  
same behaviors as electron transport

# Hole energies

# Holes



We can think of the hole kinetic energy as being positive if we look at the band diagram upside down just as we can similarly think of positive hole energies in the hole Fermi-Dirac distribution when looking at the band diagram upside down

# Holes

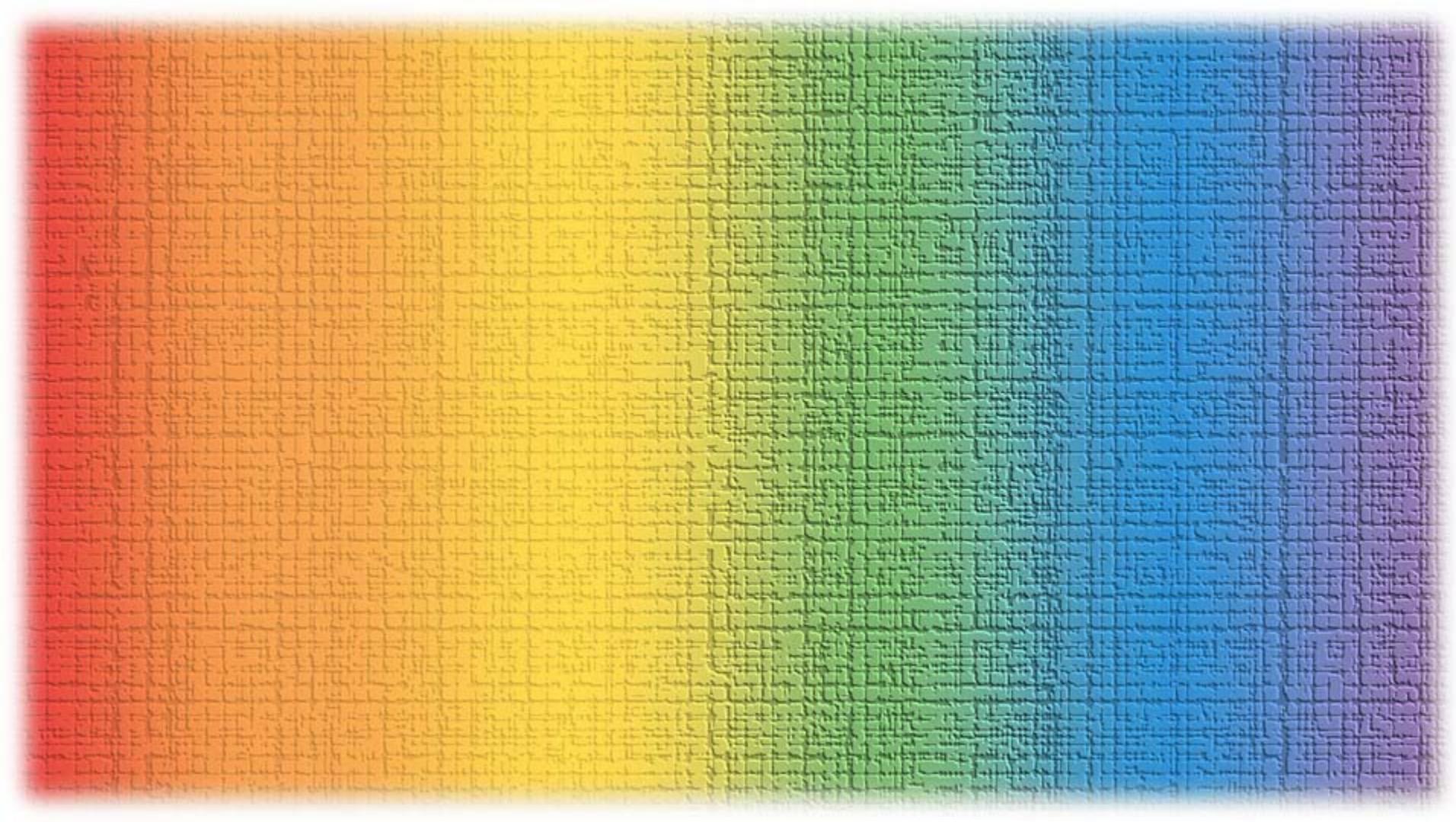


Generally, for holes

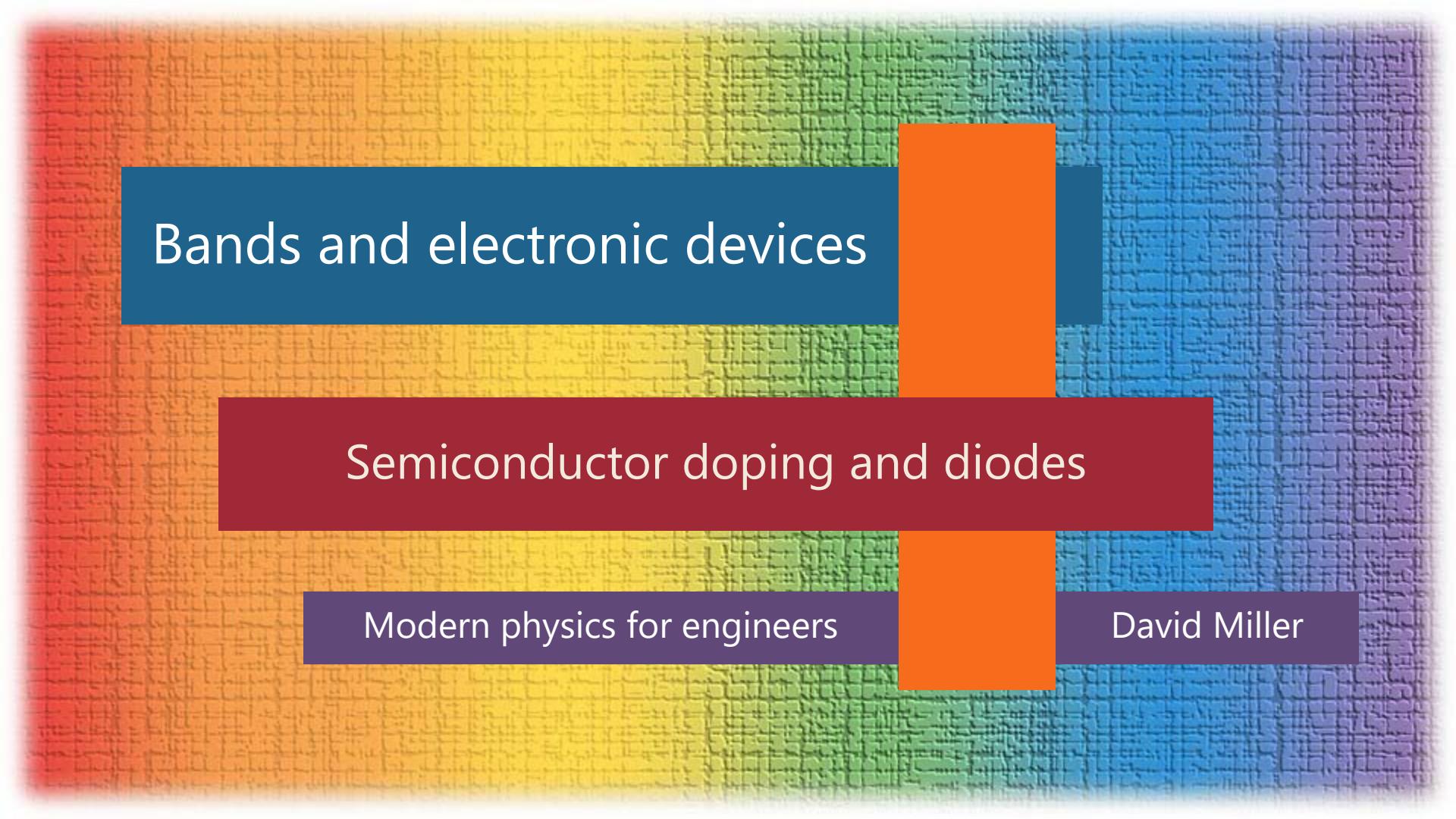
we "stand on our heads" in looking  
at energy diagrams

just as we would for thinking  
about

the energy of bubbles in a liquid  
which is a higher energy for a  
bubble "deeper" in the liquid







# Bands and electronic devices

## Semiconductor doping and diodes

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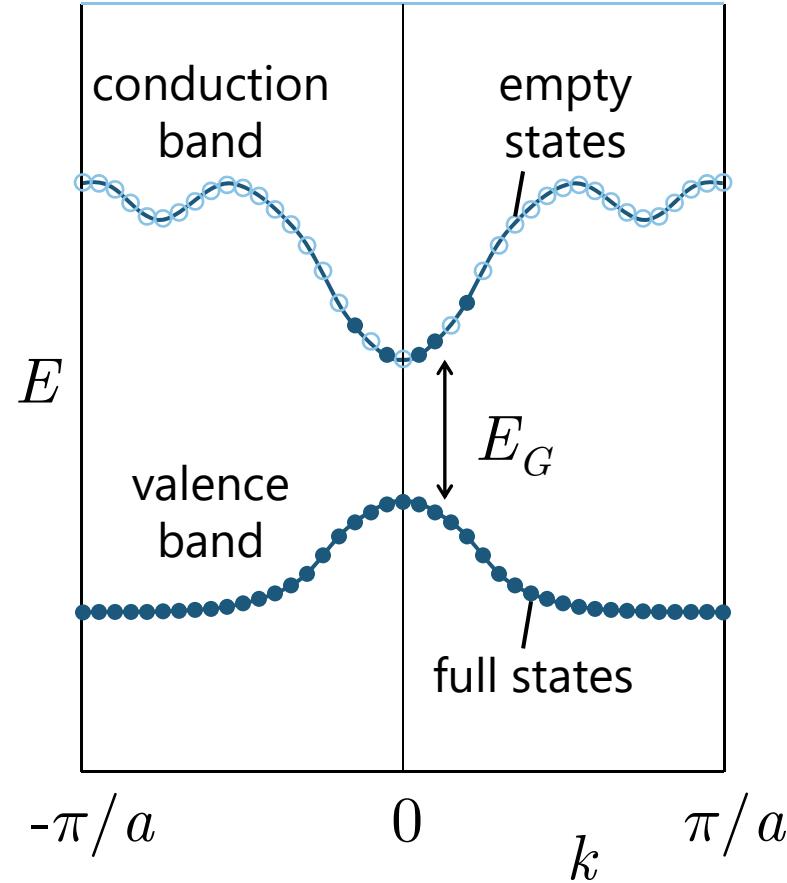
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# Doping in semiconductors

# Doping semiconductors

Substituting a few atoms with more electrons

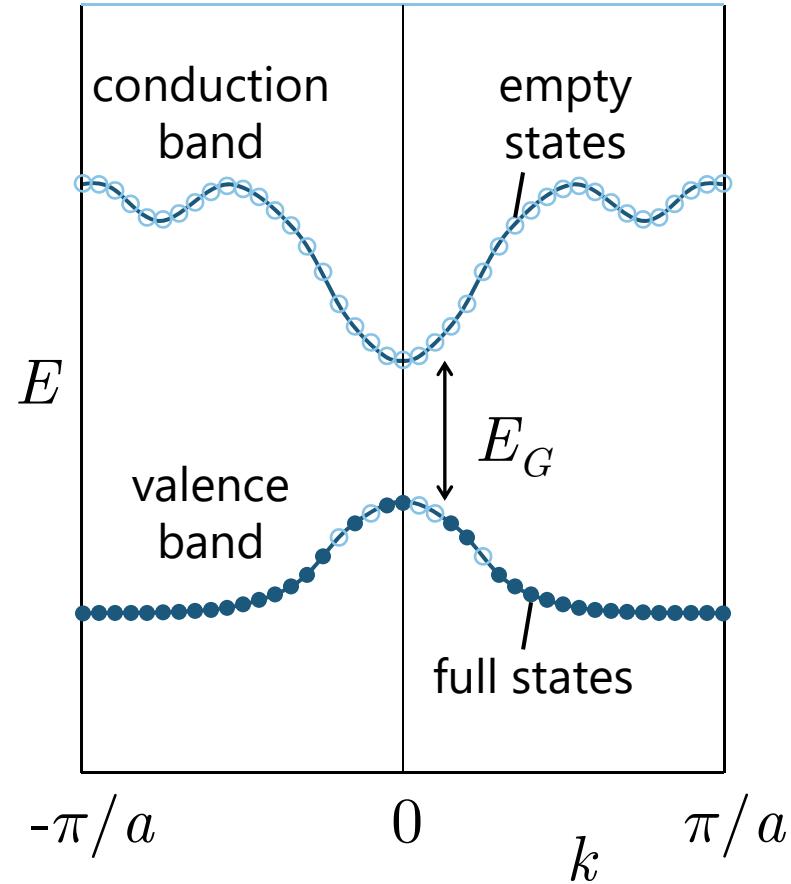
e.g., a Group V element like phosphorus in a Group IV semiconductor like silicon known as n-type doping makes the material conduct more using these additional electrons



# Doping semiconductors

Substituting a few atoms with fewer electrons

e.g., a Group III element like boron in a Group IV semiconductor like silicon known as p-type doping makes the material conduct more using these additional "holes"



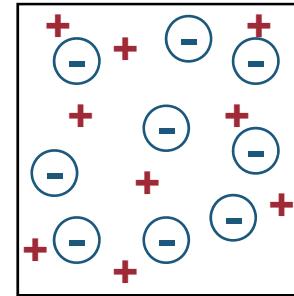
# Semiconductor diodes

# Semiconductor diode

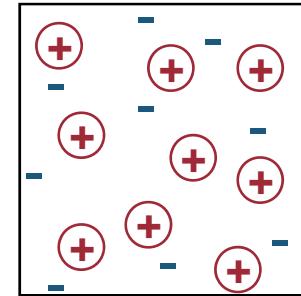
Conceptually, to make a diode  
we join

a piece of p-doped  
semiconductor  
with ionized acceptors  
and free holes

to  
a piece of n-doped  
semiconductor  
with ionized donors  
and free electrons



*p*-type



*n*-type

⊖ ionized acceptor

⊕ free hole

⊕ ionized donor

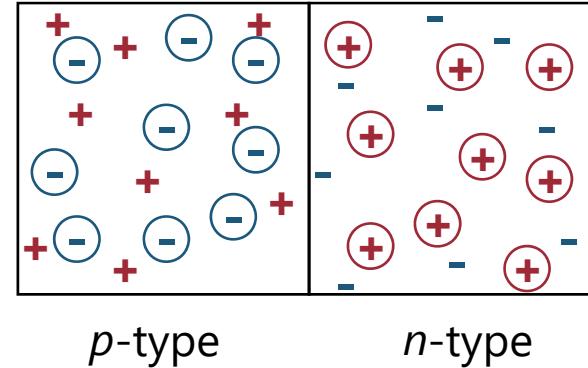
⊖ free electron

# Semiconductor diode

Conceptually, to make a diode  
we join

a piece of p-doped  
semiconductor  
with ionized acceptors  
and free holes

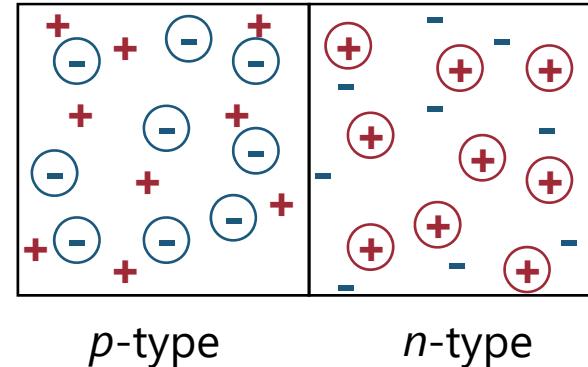
to  
a piece of n-doped  
semiconductor  
with ionized donors  
and free electrons



⊖ ionized acceptor                            ⊕ ionized donor  
+ free hole                                    - free electron

# Semiconductor diode

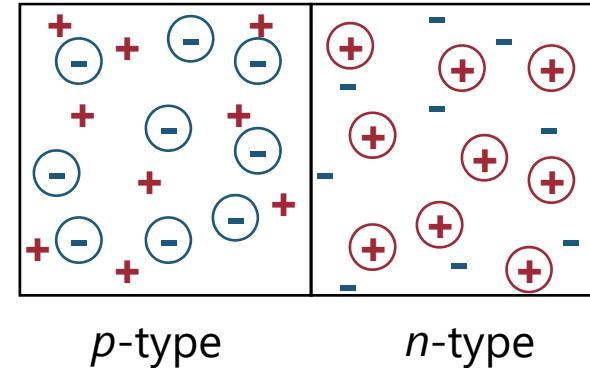
Once they are joined  
free electrons move  
from the side with more  
to the side with less  
by “diffusion”  
and similarly for free holes



- ⊖ ionized acceptor
- ⊕ ionized donor
- ⊕ free hole
- ⊖ free electron

# Semiconductor diode

Diffusion is the process where  
as a result of a “random walk”  
there is net flow  
from regions of high  
concentration  
to regions of low  
concentration  
like smoke diffusing through  
a room

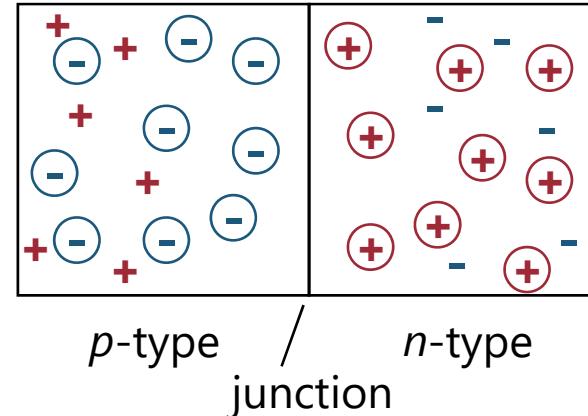


(-) ionized acceptor      (+) ionized donor  
+ free hole      - free electron

# Semiconductor diode

Free electrons and free holes

arriving in the same region as  
a result of diffusion  
effectively “annihilate” one  
another by recombination



⊖ ionized acceptor

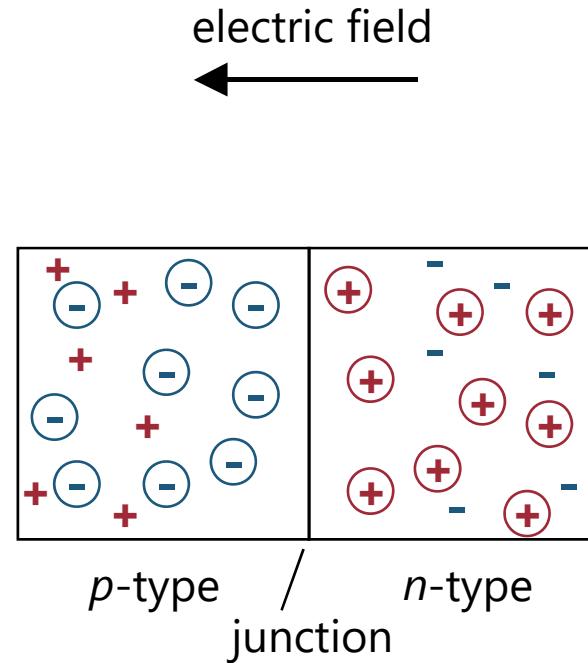
⊕ free hole

⊕ ionized donor

⊖ free electron

# Semiconductor diode

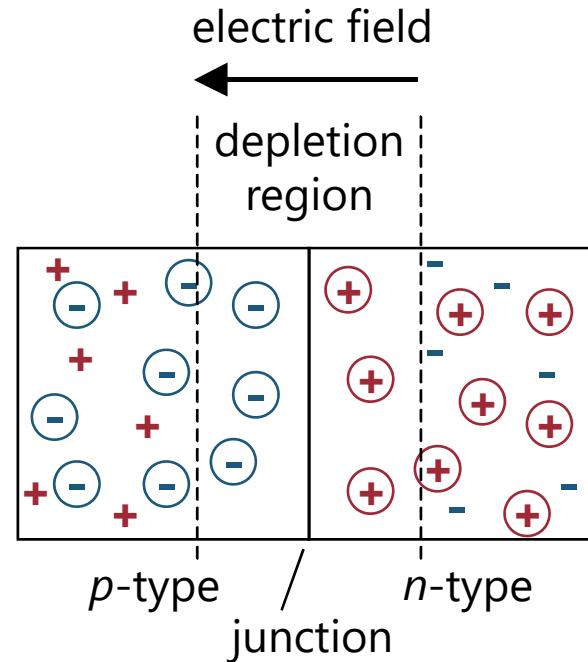
As the electrons and holes move and “annihilate” they leave behind net “bare” fixed charges the ionized donors and acceptors which means an electric field is generated in the direction that opposes the diffusion



- ionized acceptor
- +
- free hole
- +
- ionized donor
- 
- free electron

# Semiconductor diode

The result is to create a “depletion region” on either side of the junction with essentially no free charges in it and an electric field

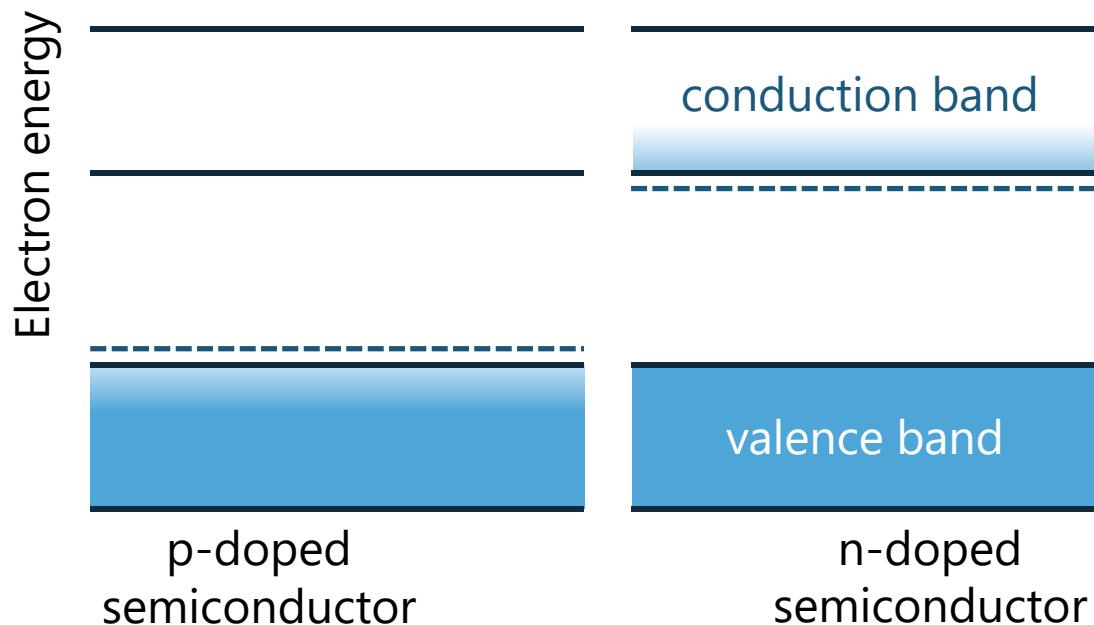


- (-) ionized acceptor      (+) free hole
- (+) ionized donor      (-) free electron

# Semiconductor diode

In terms of bands

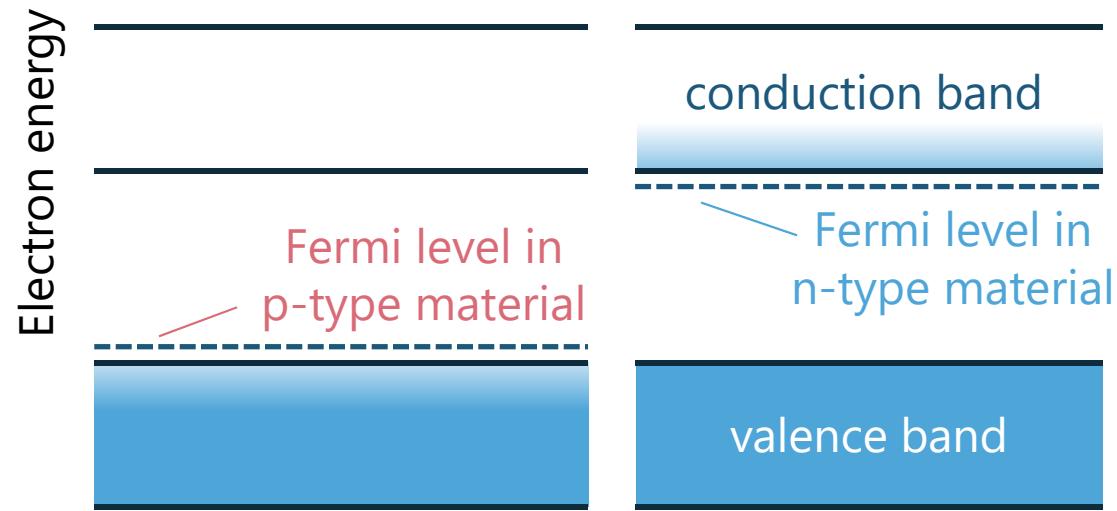
we might view the  
bands as being  
“lined up” before this  
diffusion



# Semiconductor diode

Before diffusion, the Fermi level must be near the conduction band edge in n-type material to have many electrons in the conduction band

near the valence band edge in p-type material to have many holes in the valence band



# Semiconductor diode

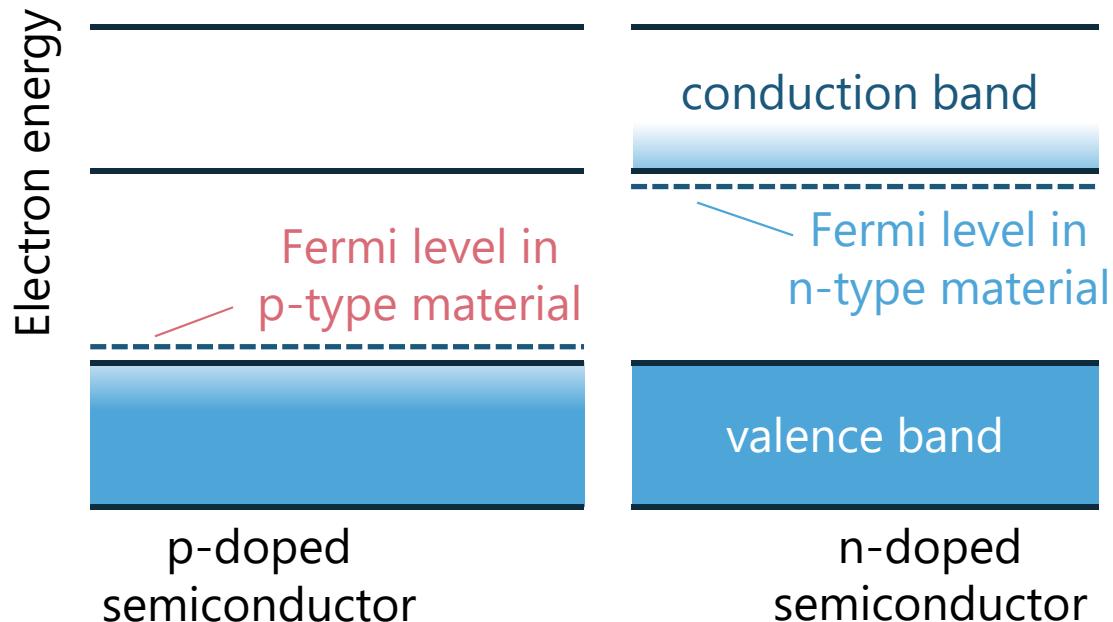
Note that we move the Fermi level

closer to the valence band

by adding p-type dopants

closer to the conduction band

by adding n-type dopants



# Semiconductor diode

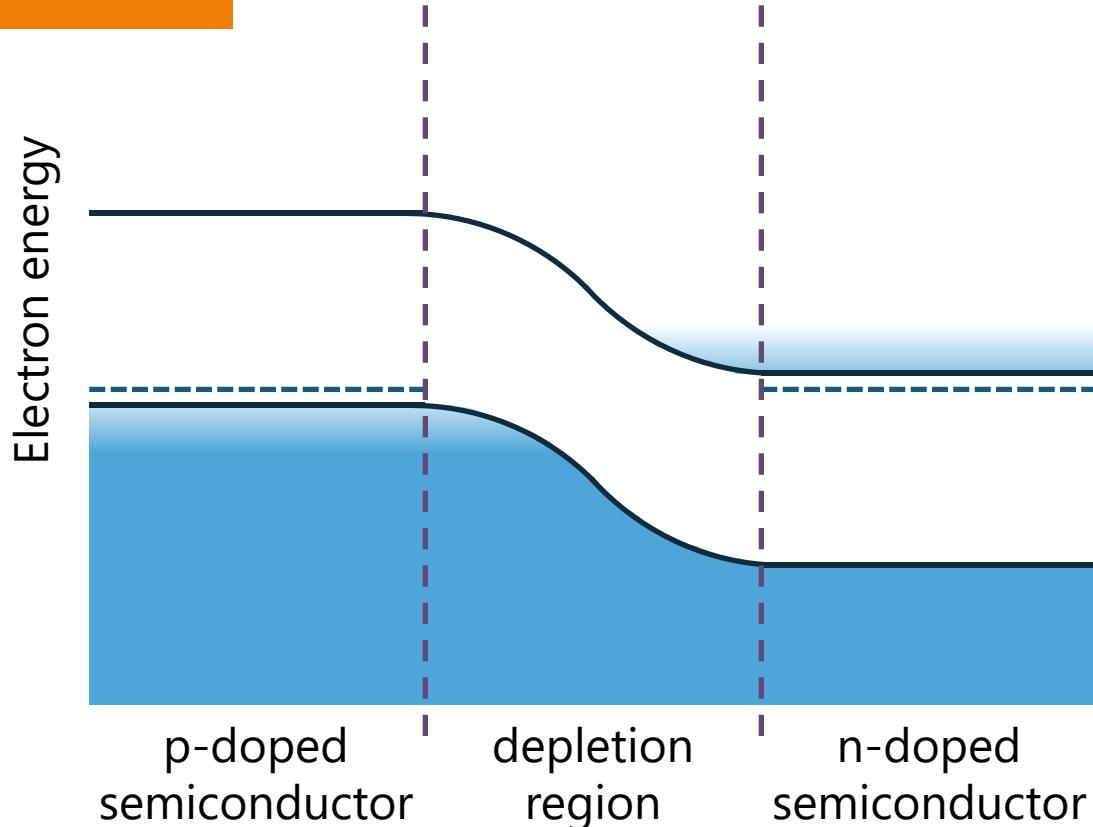
After the diffusion

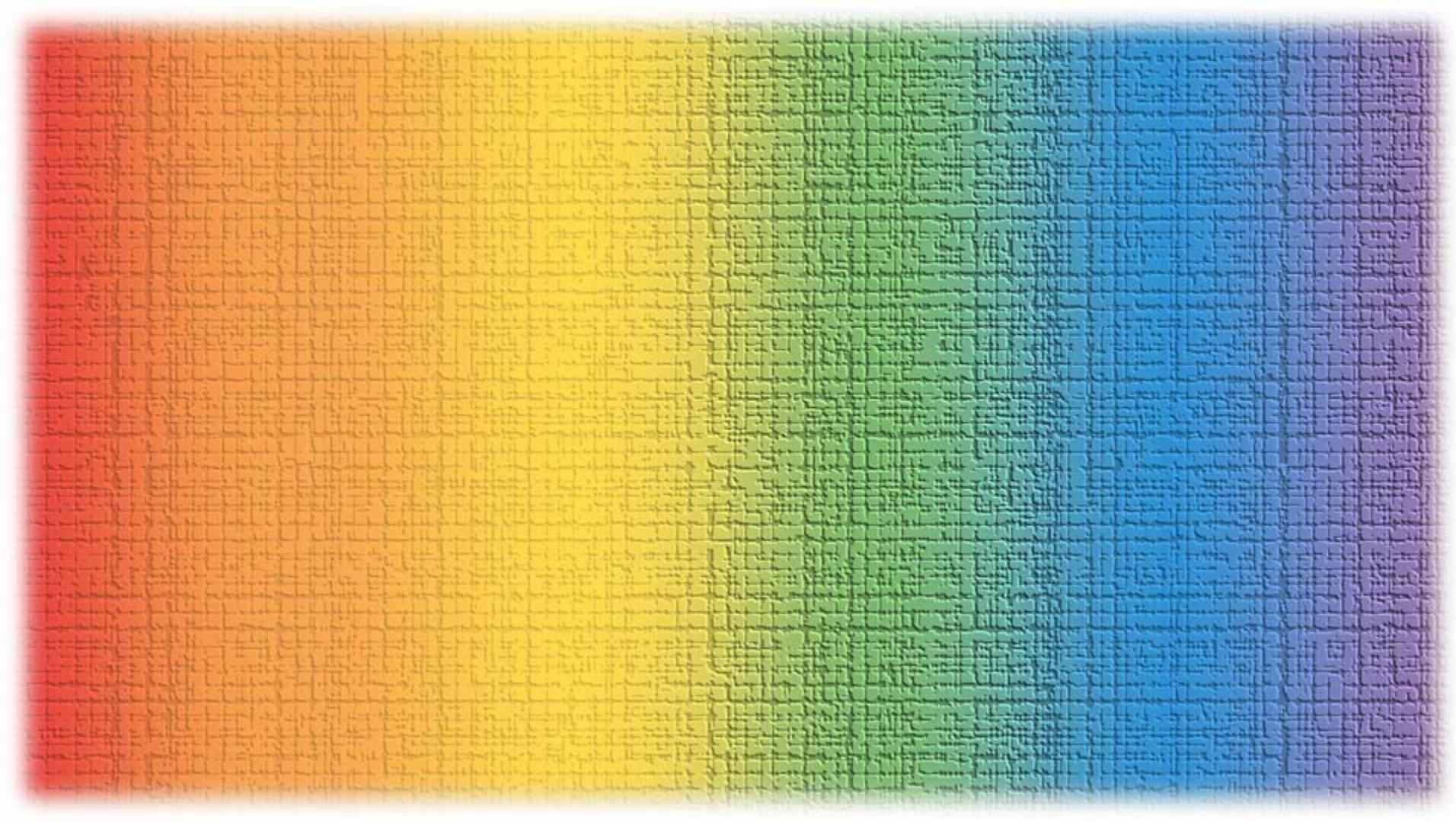
the electrostatic potential  
has  
“bent” the bands

Formally

once the diffusion is  
finished

the “Fermi levels” or  
“chemical potentials”  
are equalized







# Bands and electronic devices

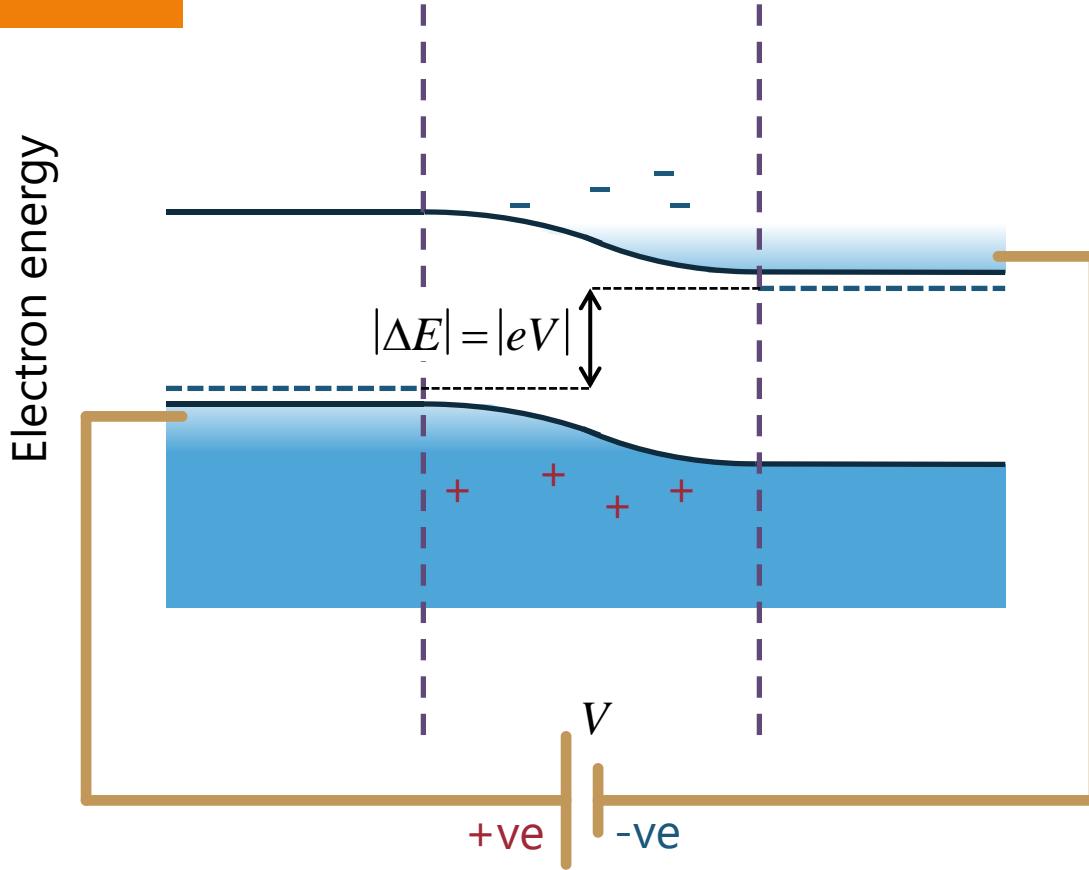
Voltages and Fermi levels

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# Voltages and Fermi levels

Why is it that  
when we apply a  
voltage  $V$  between the  
two sides of a diode  
we separate the  
Fermi levels  
by an amount  
 $|\Delta E| = |eV|$  ?

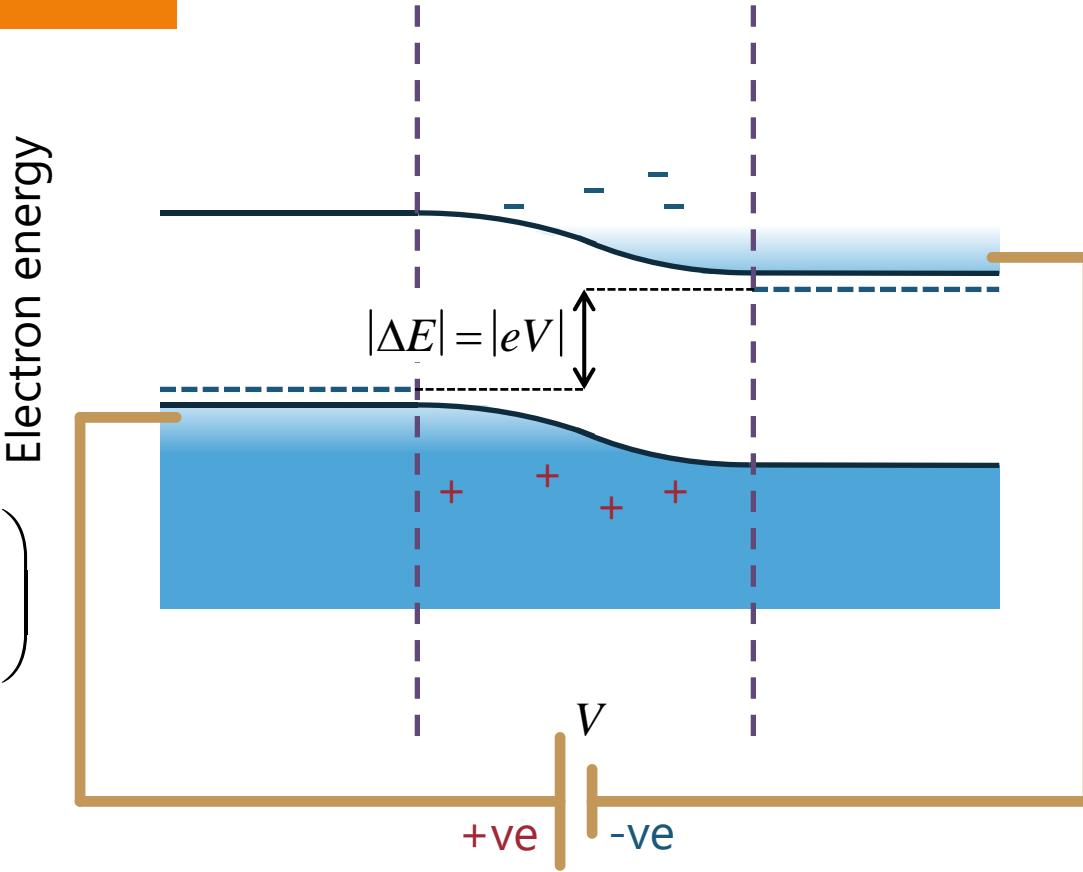


# Voltages and Fermi levels

To understand this  
we need an alternate  
but equivalent  
definition of chemical  
potential

$$\mu_c = \left( \frac{\partial U}{\partial N} \right) \bigg|_S \quad \left( \equiv -\tau \left( \frac{\partial \sigma}{\partial N} \right) \bigg|_U \right)$$

where  $U$  is energy

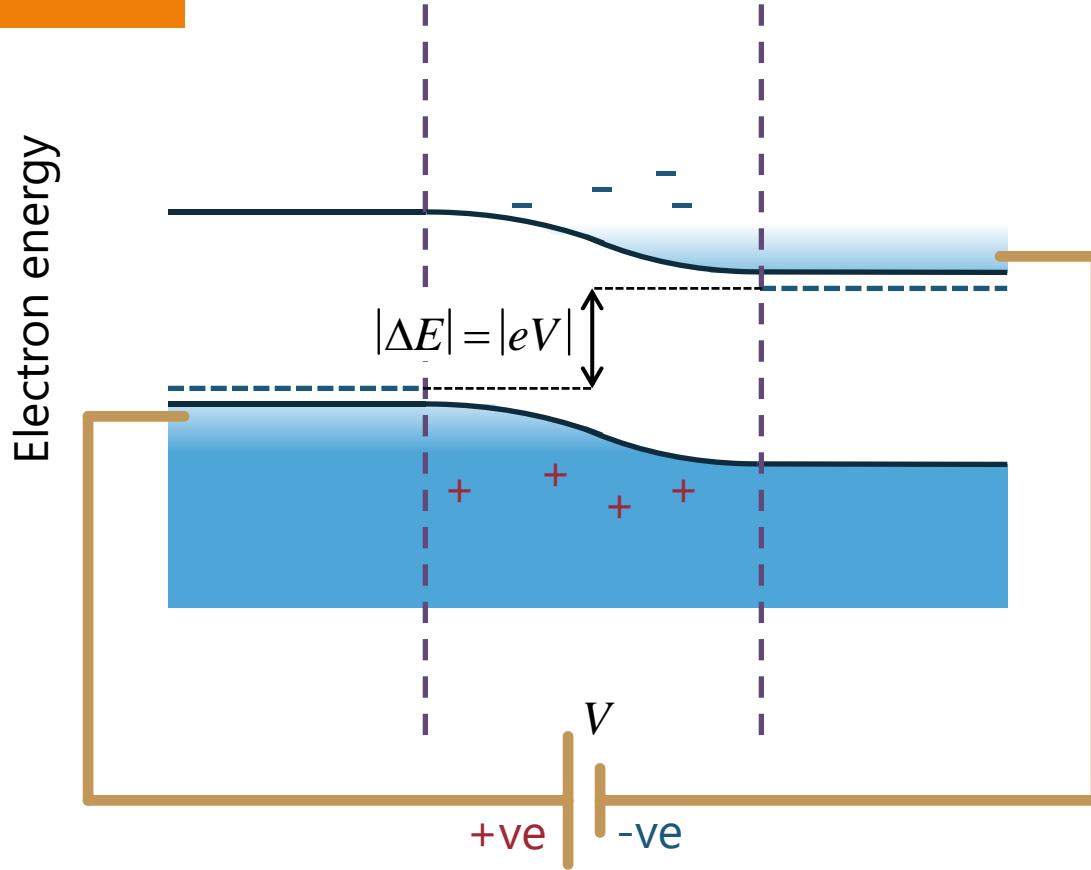


# Voltages and Fermi levels

First, we can understand why this alternative form

$$\mu_c = \left( \frac{\partial U}{\partial N} \right) \Big|_s$$

gives the result we need



# Voltages and Fermi levels

Note that  $\left(\frac{\partial U}{\partial N}\right)_{\!S}$  is

the energy per particle at constant entropy

Adding a potential energy  $|\Delta E| = |eV|$  to every electron  
by changing the voltage  $V$

changes the energy per particle by  $|\Delta E| = |eV|$

without changing the entropy at all

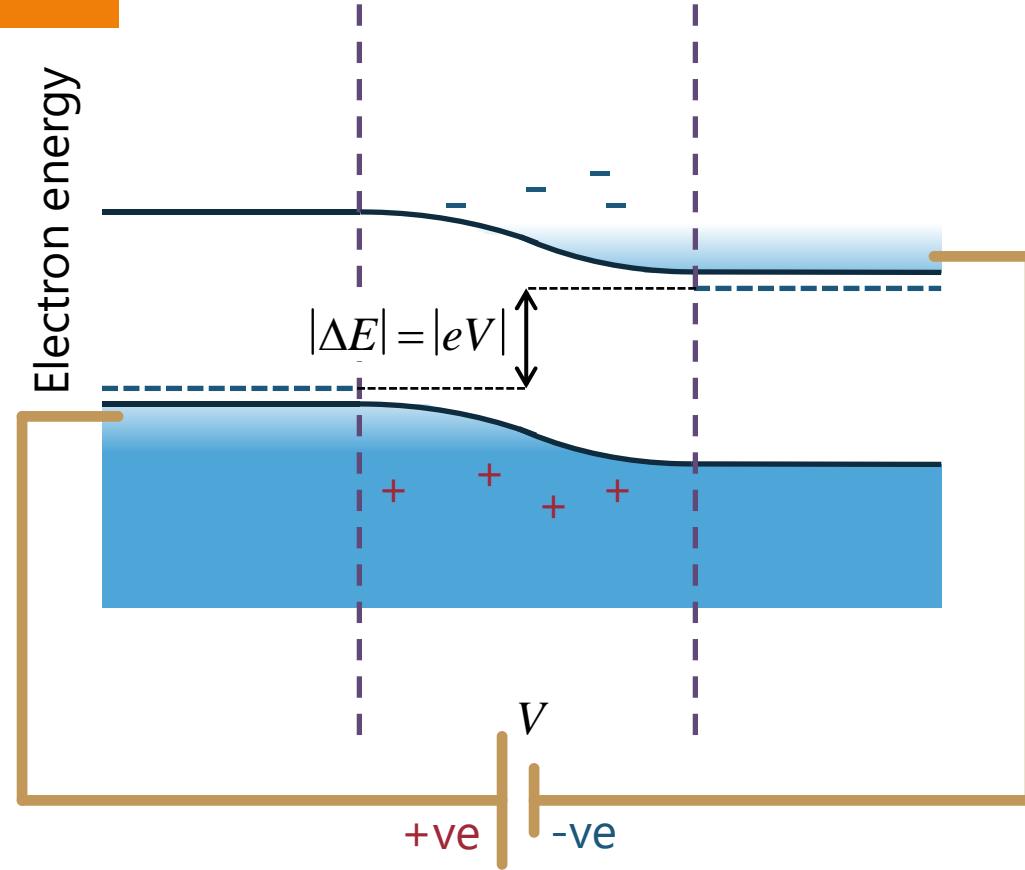
and hence changes the chemical potential

$$\mu_c = \left(\frac{\partial U}{\partial N}\right)_{\!S} \text{ by the same amount}$$

Q.E.D

# Voltages and Fermi levels

Hence adding  $|\Delta E| = |eV|$  to each electron on the "right" compared to the electrons on the "left" raises the Fermi level (chemical potential) on the "right" by  $|\Delta E| = |eV|$  compared to that on the "left"



# Proof of equivalence of chemical potential definitions

Consider the differential of entropy  $d\sigma$  for a system with energy  $U$  and particle number  $N$

Then, considering a situation of constant entropy, we would have

$$d\sigma = \left( \frac{\partial \sigma}{\partial U} \right)_{\!N} dU + \left( \frac{\partial \sigma}{\partial N} \right)_{\!U} dN = 0$$

So, at such a constant entropy, we have

$$\left( \frac{\partial \sigma}{\partial U} \right)_{\!N} dU = - \left( \frac{\partial \sigma}{\partial N} \right)_{\!U} dN$$

# Proof of equivalence of chemical potential definitions

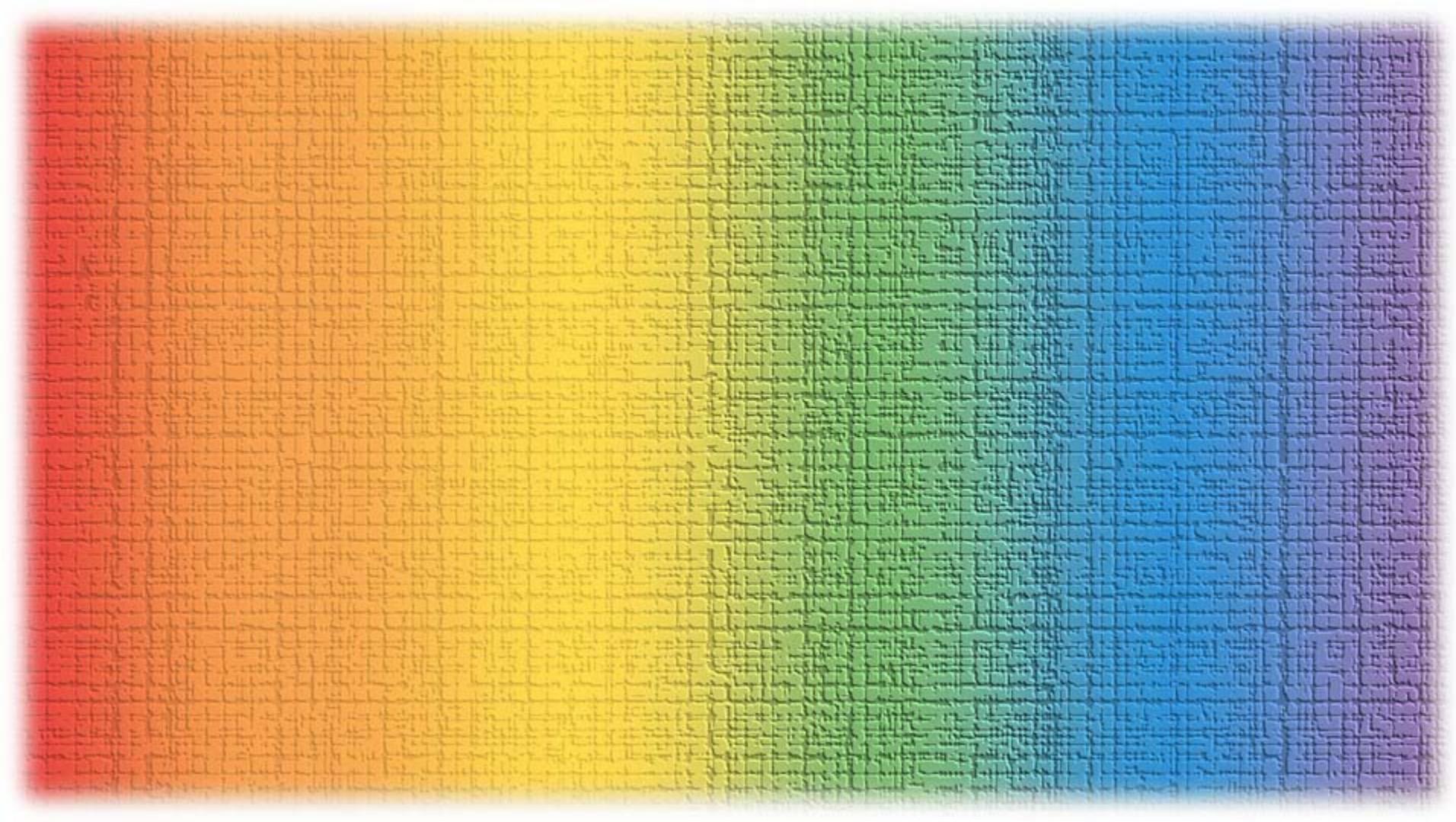
With  $\left(\frac{\partial\sigma}{\partial U}\right)_N dU = -\left(\frac{\partial\sigma}{\partial N}\right)_U dN$  at constant entropy

we can therefore write  $\left(\frac{\partial\sigma}{\partial U}\right)_N \left(\frac{\partial U}{\partial N}\right)_S = -\left(\frac{\partial\sigma}{\partial N}\right)_U$

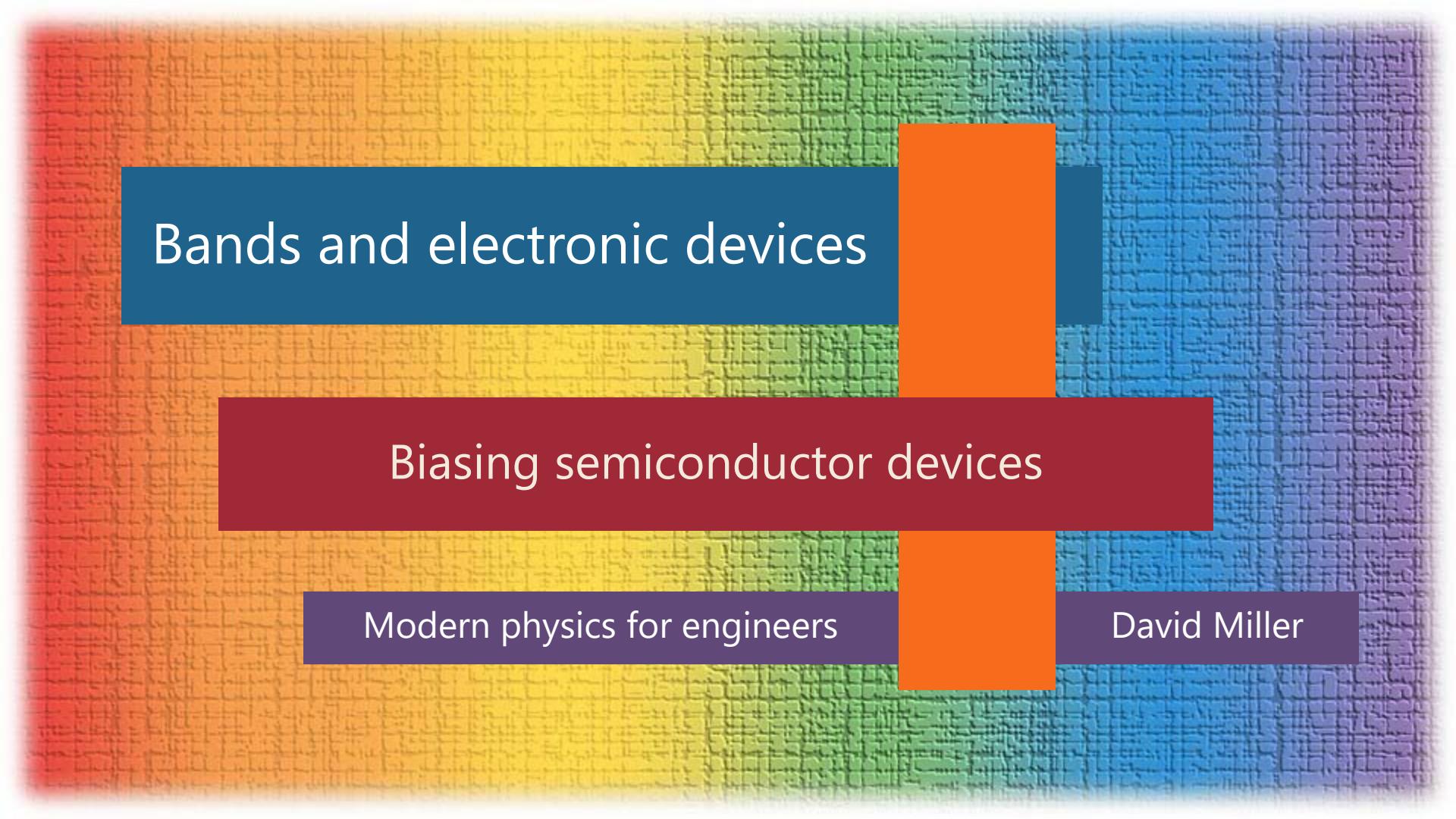
But  $\left(\frac{\partial\sigma}{\partial U}\right)_N \equiv \frac{1}{\tau}$

So  $\left(\frac{\partial U}{\partial N}\right)_S = -\tau \left(\frac{\partial\sigma}{\partial N}\right)_U = \mu_C$

which is the chemical potential – Q. E. D.







# Bands and electronic devices

## Biassing semiconductor devices

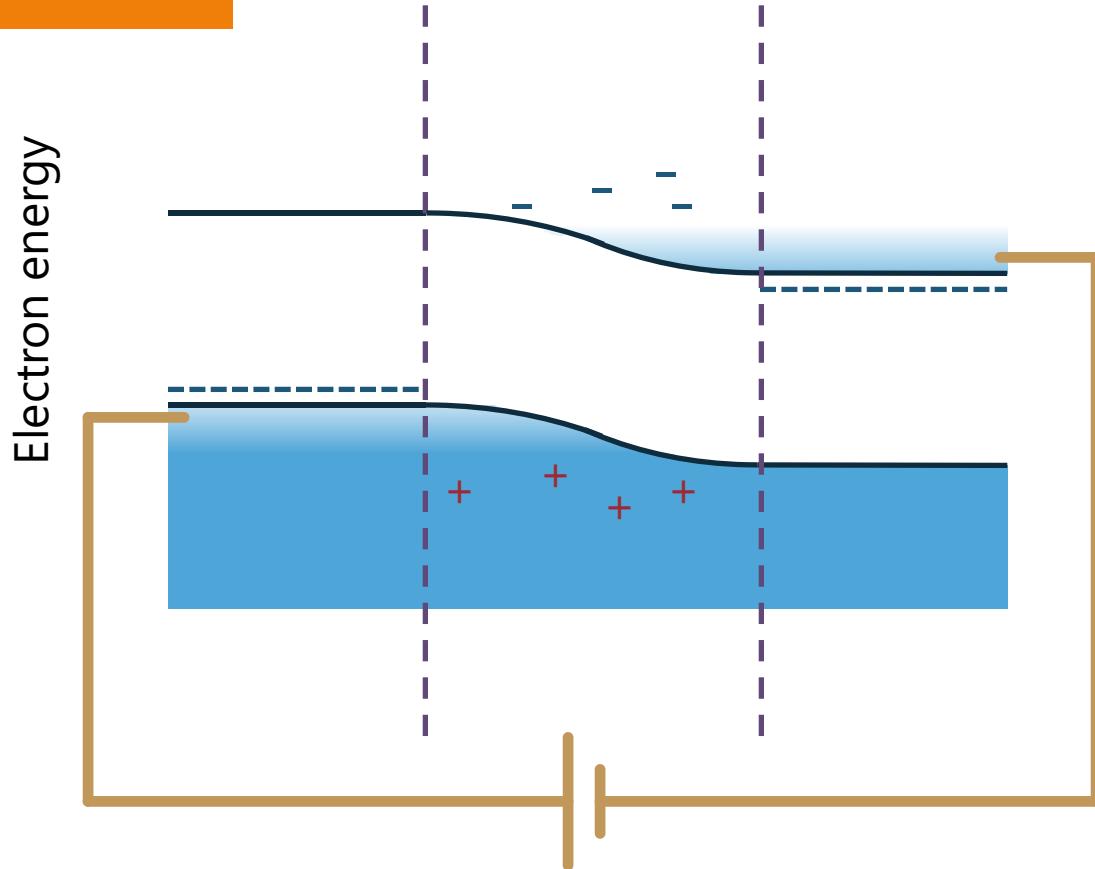
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## Biassing semiconductor diodes

# Semiconductor diode

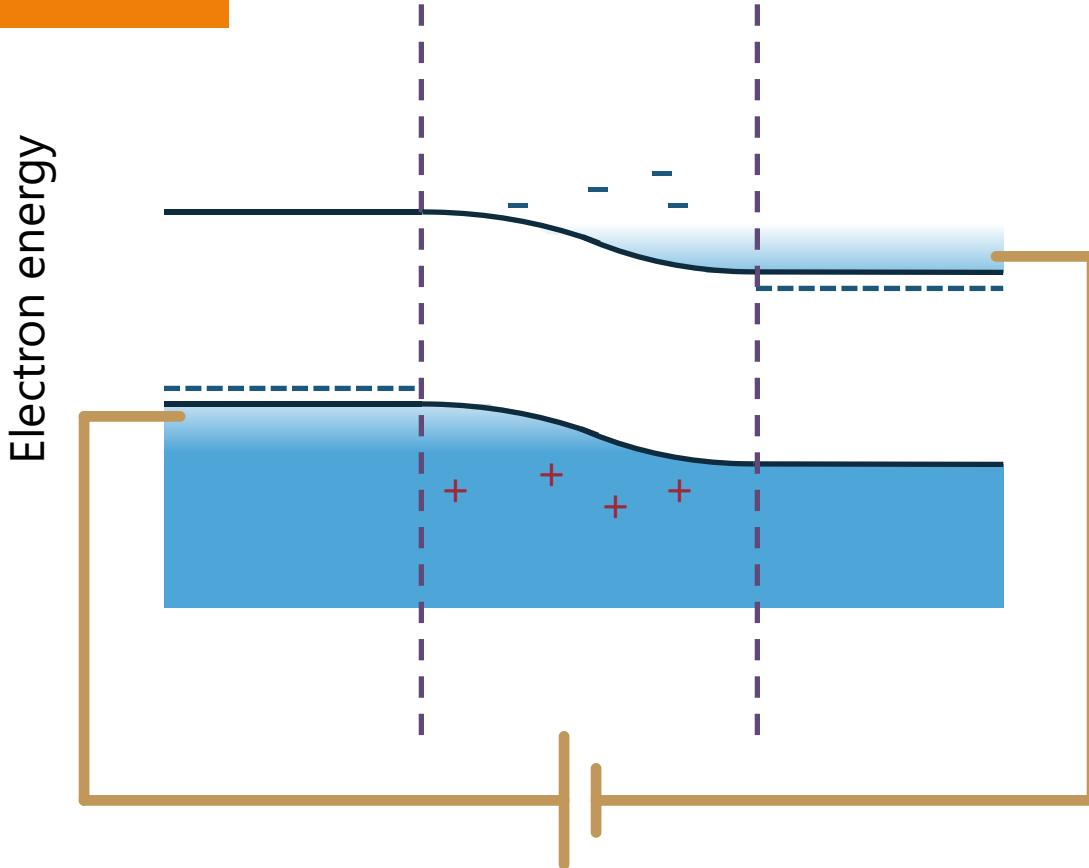
“Forward biasing”  
“unbalances” the  
diffusion again  
allowing electrons  
with enough  
thermal energy  
to diffuse from  
right to left  
over the  
potential



# Semiconductor diode

and allowing holes with enough thermal energy to diffuse from left to right “over” the potential

This diffusion current is the normal diode forward current



# Reverse-biased diode

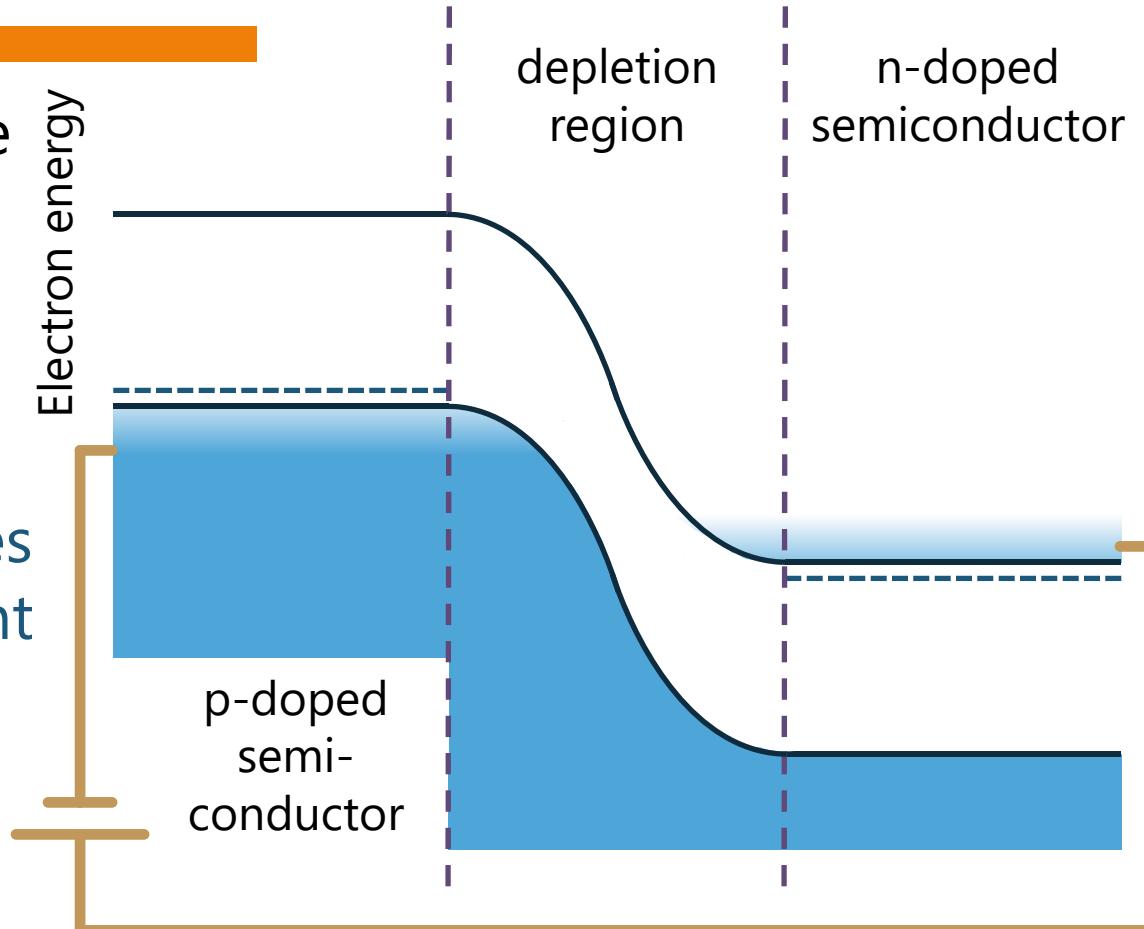
In a reverse-biased diode

the barriers

for conduction band electrons to diffuse to the left, and for valence band holes to diffuse to the right

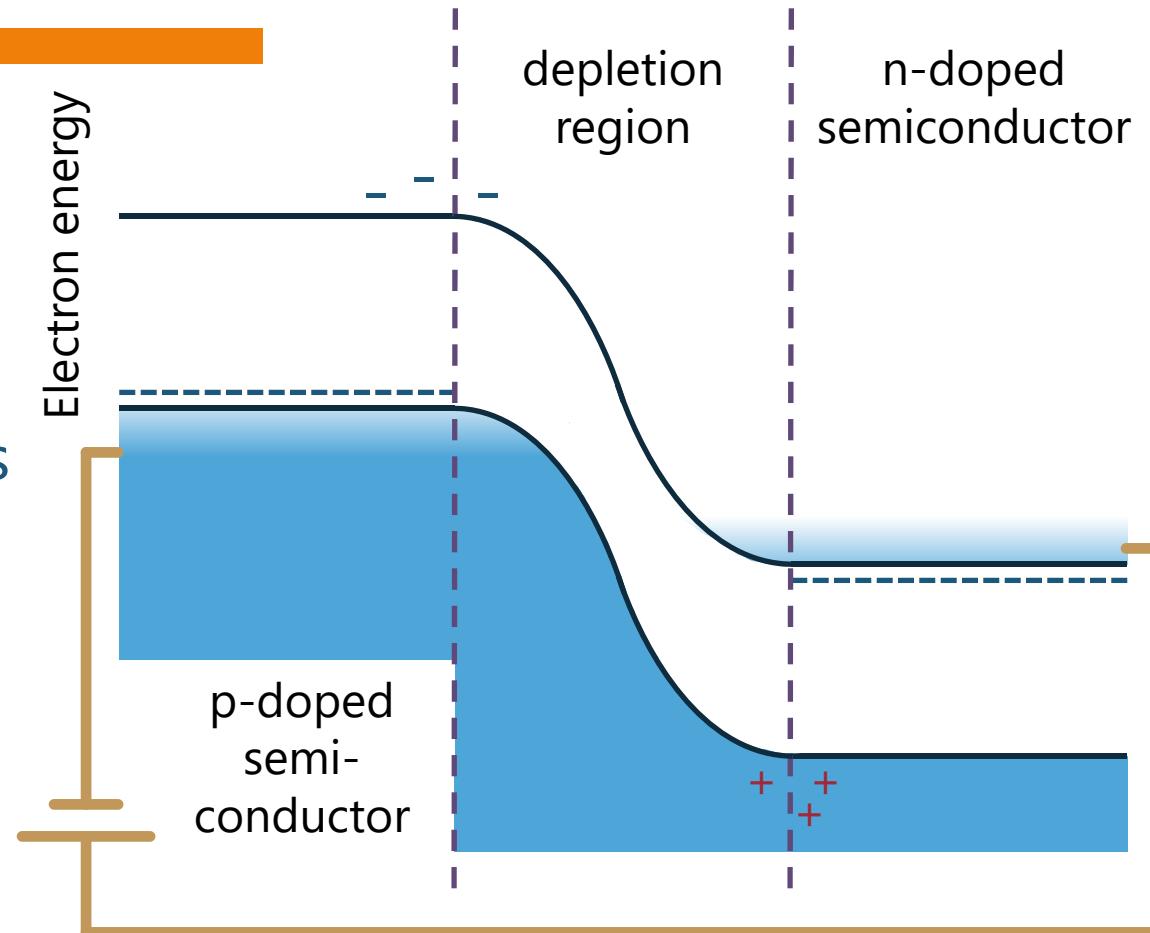
are even higher

turning off forward diffusion current



# Reverse-biased diode

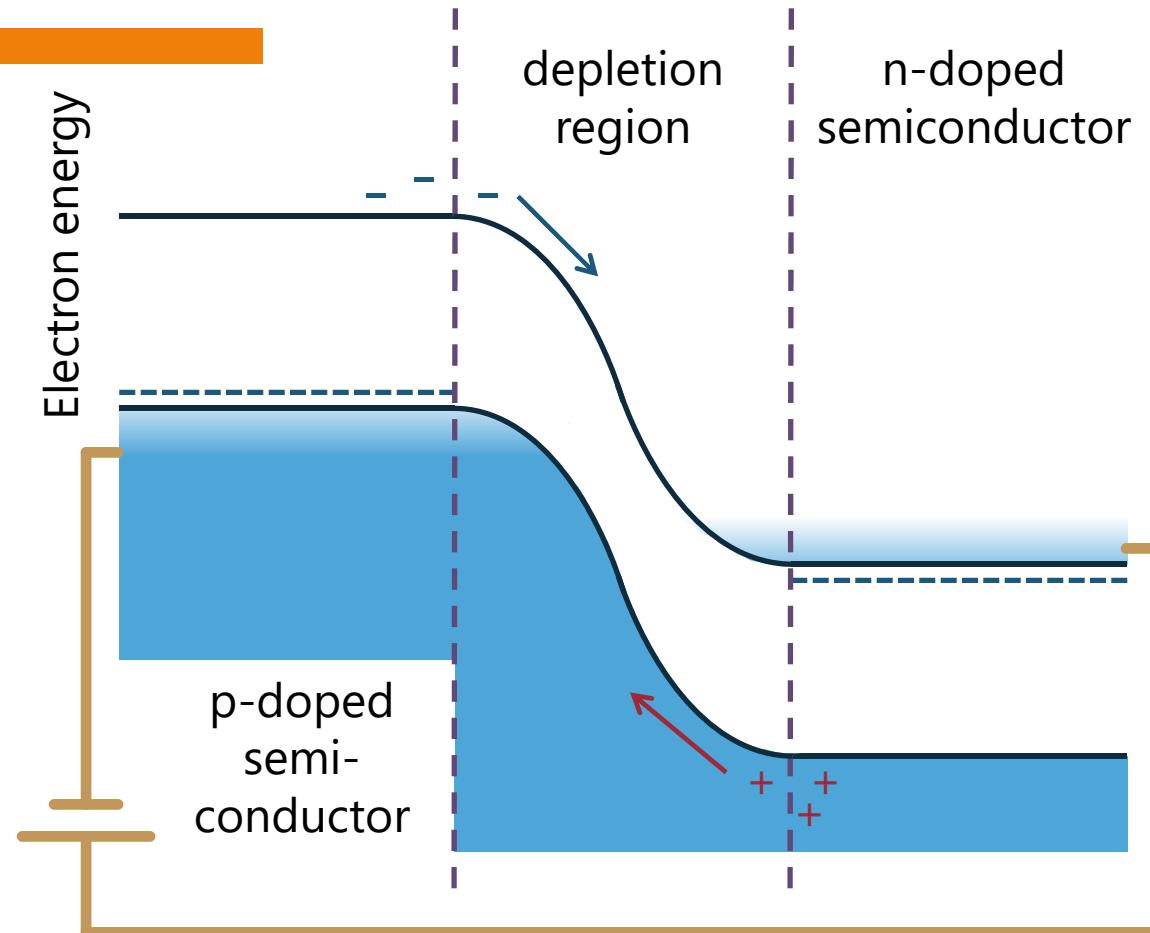
At a finite temperature there are also very small “minority carrier” densities of conduction electrons in the p-semiconductor and valence holes in the n-semiconductor



# Reverse-biased diode

These minority carriers  
can diffuse into the  
depletion region  
and drift down-hill

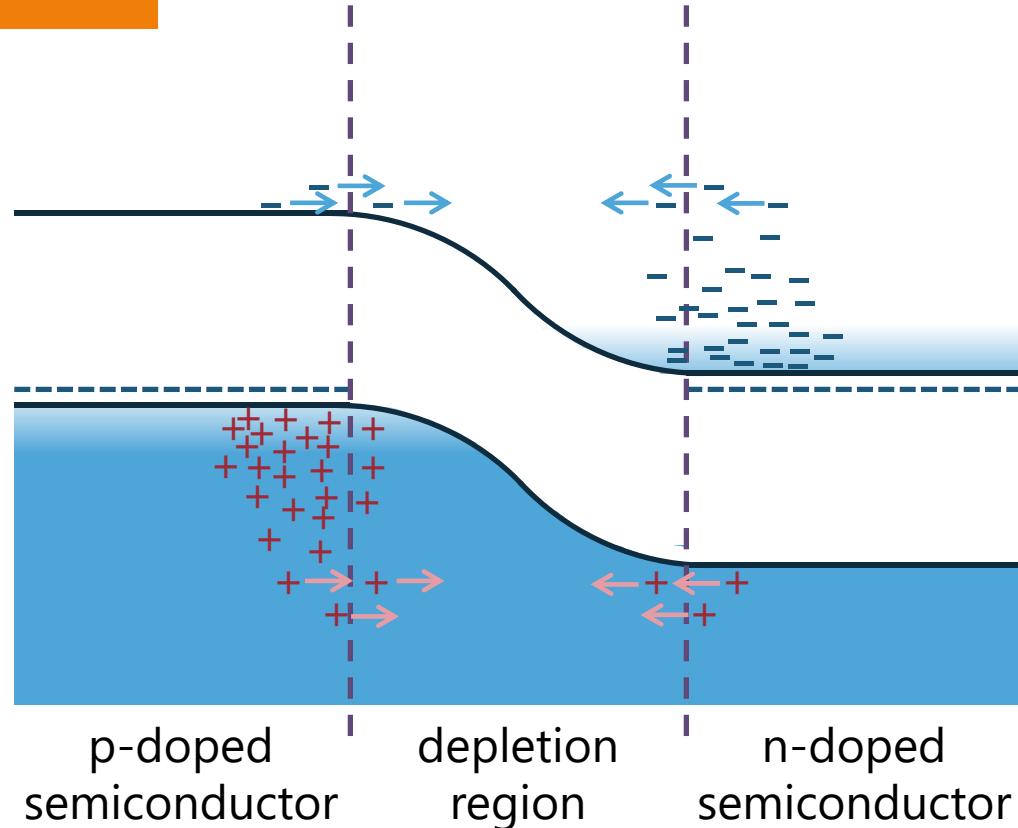
Under reverse bias  
this gives reverse  
leakage current  
of magnitude  $I_S$   
present even in an  
ideal diode



# Semiconductor diode current-voltage characteristic

# Current at zero bias voltage

At zero bias voltage  
the number of  
minority carriers  
diffusing in "reverse"  
equals the number of  
"majority carriers"  
diffusing "forwards"  
electrons on the n  
side  
holes in the p side

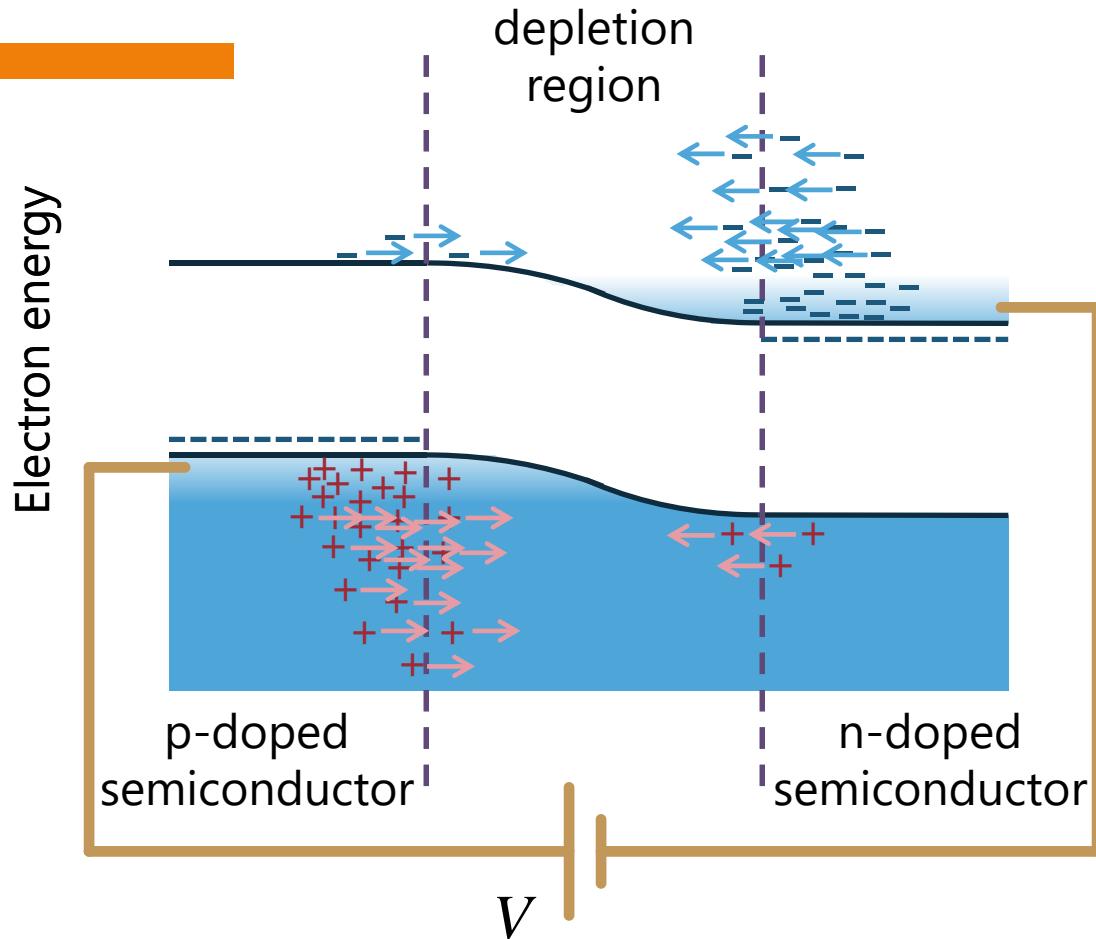


# Net forward current

Under forward bias by  $V$  volts

the occupation probability of the majority carrier states presuming the Maxwell-Boltzmann approximation has increased by

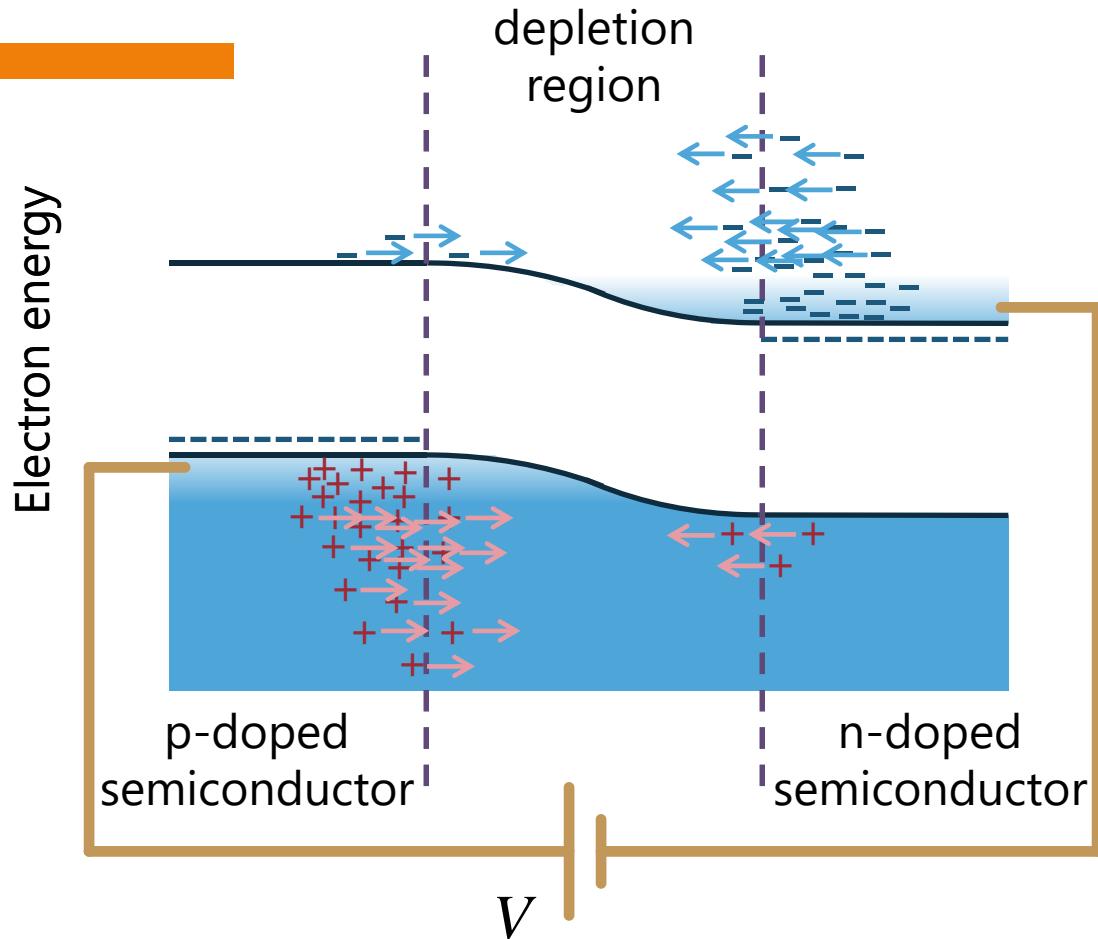
$$\exp(eV / k_B T)$$



# Net forward current

So the net forward current  $I$  in a diode is the forward diffusion minus the backward "leakage" diffusion

$$I = I_s \left[ \exp\left(\frac{eV}{k_B T}\right) - 1 \right]$$



# Diode current-voltage (I-V)

Here is current as a function of voltage for an “ideal” diode

At room temperature

$$k_B T / e \simeq 25 \text{ mV}$$

$I_S$  depends on doping levels and material properties

$$I = I_S \left[ \exp\left(\frac{eV}{k_B T}\right) - 1 \right]$$

